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# Introduction

Recently, the field of research has been led towards the production of innovative and inexpensive materials able to degrade emerging compounds, such as persistent organic pollutants (POPs). In this regard, heterogeneous photocatalysis offers a green solution for water remediation [1].

Furthermore, a photocatalytic floating device can be adopted to overcome issues related to the use of catalystbased slurry systems and to maximize light utilization. In the present work, photoactive materials were prepared immobilizing visible light responsive catalysts on floating supports (synthetic and natural polymers, inorganic materials) and tested towards the degradation of a model molecule (Rhodamine B RHB).



# LECA

- Advantages:
- Cheap material
- Eco-compatible
- Stress-resistance

#### Disadvantages

- Instability of the coating
- Inertness of the material



# **Innovative Floating Materials For Water Remediation** Under Sunlight

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**impregnation** [4]

### Results



best materials have been subjected to recycle tests. It was observed that high maintain

**Conclusions:** All the mentioned materials were promising for the development of floating photocatalysts in order to photo-degrade organic pollutants in water. However, further studies shall be carried out to find a compromise between coating stability/activity and eco-compatibility.

# Materials & Methods





Grafitic carbon nitride on Calcium alginate Spheres (Alg) via **ionotropic gelation** [3]

#### Advantages: 100 Light 90 - Cheap material 80 - Easy synthetic route 60 - Eco-compatible 50 40 Disadvantages 20 - Deactivation catalysts of 10 $(TiO_2)$ caused by carbonates instability of - Long-term calcium alginate

[4] Y. Cao et al,. Synt. Met., **1995**, *69*, 187-190. Acknowledgments: the authors thank *Velux Stiftung Foundation* for the financial support through the project 1381 "SUNFLOAT-Water decontamination by sunlight-driven floating photocatalytic systems".



Floating photocatalysts activity was assessed through the photodegradation of 10 ppm aqueous solution of rhodamine B (RHB) under solar light. Experiments last for 210 min, 30 min of dark followed by 180 min of irradiation.

Polyure			
<ul><li>PANI1(C</li><li>→PANI2(C</li></ul>	CSA)-TiC CSA)-TiC	)2_PU )2_PU	
0.0	120	1.50	100
90 ne (min)	120	150	180

# ethane

#### **Advantages**:

- Cheap material
- 2 synthetic routes: traditional PANI (PANI1) vs green PANI (PANI2)

#### Disadvantages

- PU is not eco-friendly
- PU is not suitable for the of inorganic attachment catalysts

#### **Bibliography:**

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- [2] Y. Shavisi et al., J. Ind. Eng. Chem. **2014**, *20*, 2806–2813.
- [3] I. Dalponte et al., Int. J. Bio. Macrom., **2019**, *137*, 992–1001.